

CATALYST

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Catalysis - Wikipedia

Catalysis is the process of increasing the rate of a chemical reaction by adding a substance known as a catalyst which is not consumed in the catalyzed reaction.

Catalyst definition, a substance that causes or accelerates a chemical reaction without itself being affected. See more.

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In biology, enzymes are protein-based catalysts in metabolism and catabolism. Biodiesel and related biofuels require processing Catalyst both inorganic and biocatalysts. Please try again later.

FormoredetailsandotherplatformsreadtheextendedInstallingCatalysts Because of this preactivation Catalyst, many catalytic reactions involve an induction period. Although catalysts are not consumed by the reaction itself, they may be inhibited, deactivated, or destroyed by secondary processes. It is then consumed as the reaction proceeds, Catalyst thus it is also a reactant.

Hybridautomobilesarepoweredbyanickel–lanthanummetalhydrideCatalyst catalyst stabilizes the transition state more than it stabilizes the starting material.